

Nanomateriali per applicazioni high tech



Prodotti e applicazioni

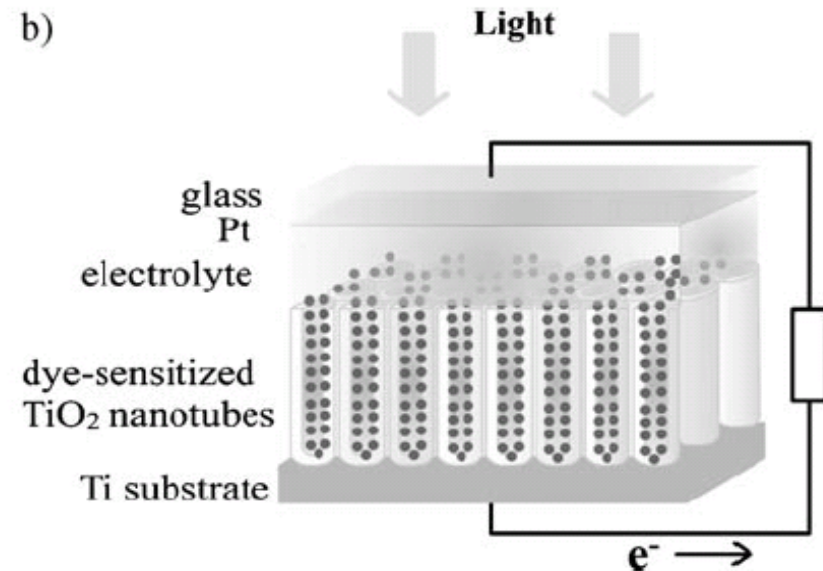
- Fotoanodi TiO_2
- Substrati colture cellulare
- Membrane PAA
- Nanofili e nanoparticelle metalliche
- Assorbitori solari

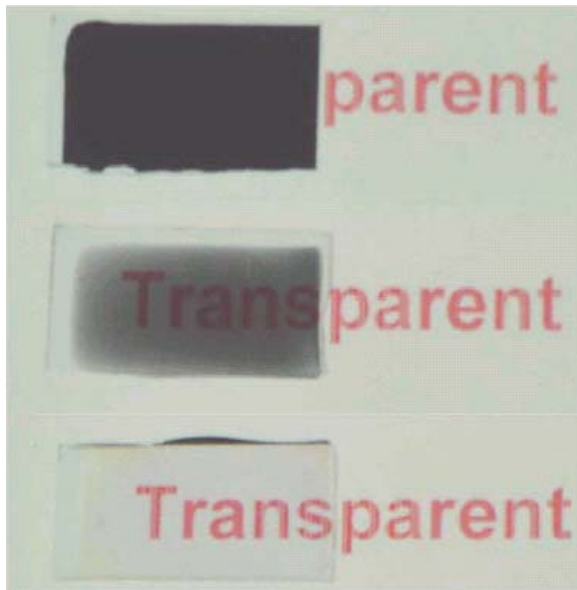
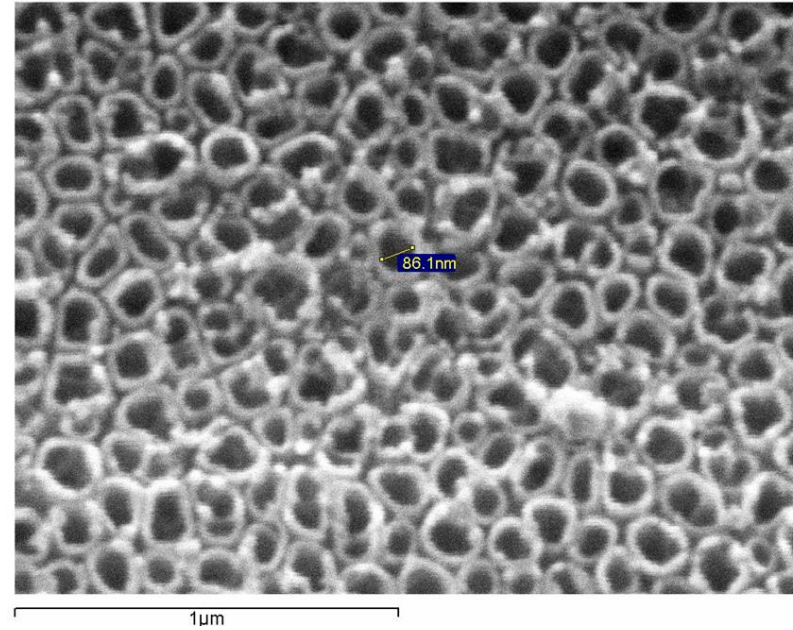
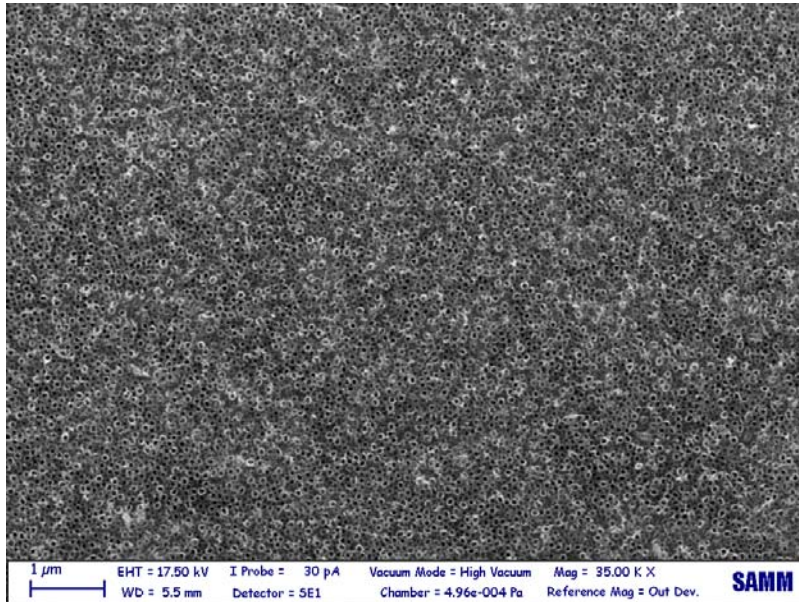


Ossido anodico nanotubolare TiO₂

Celle fotovoltaiche di terza generazione
(DSSC o celle di Grätzel)

- Elettrodo ad elevata area superficiale
- Materiale semiconduttore con grande band gap ed elevata mobilità elettronica
- Minimo numero di siti di ricombinazione
- Elettrodi nanostrutturati di TiO₂, ZnO, SnO₂, ...





Ad oggi il fotoanodo consiste di nanoparticelle sinterizzate di TiO_2 . Tuttavia è stato suggerito che la sostituzione con uno strato nanostrutturato 1D (nanofili, nanotubi, nanocolonne,...) possa migliorare le prestazioni.



Substrato per culture cellulari, cellule staminali

communications

Stem cells

TiO₂ Nanotube Surfaces: 15 nm — An Optimal Length Scale of Surface Topography for Cell Adhesion and Differentiation**

Jung Park, Sebastian Bauer, Karl Andreas Schlegel, Friedrich W. Neukam, Klaus von der Mark, and Patrik Schmuki*



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www.elsevier.com/locate/actabiomat

Brief communication

Improved attachment of mesenchymal stem cells on super-hydrophobic TiO₂ nanotubes

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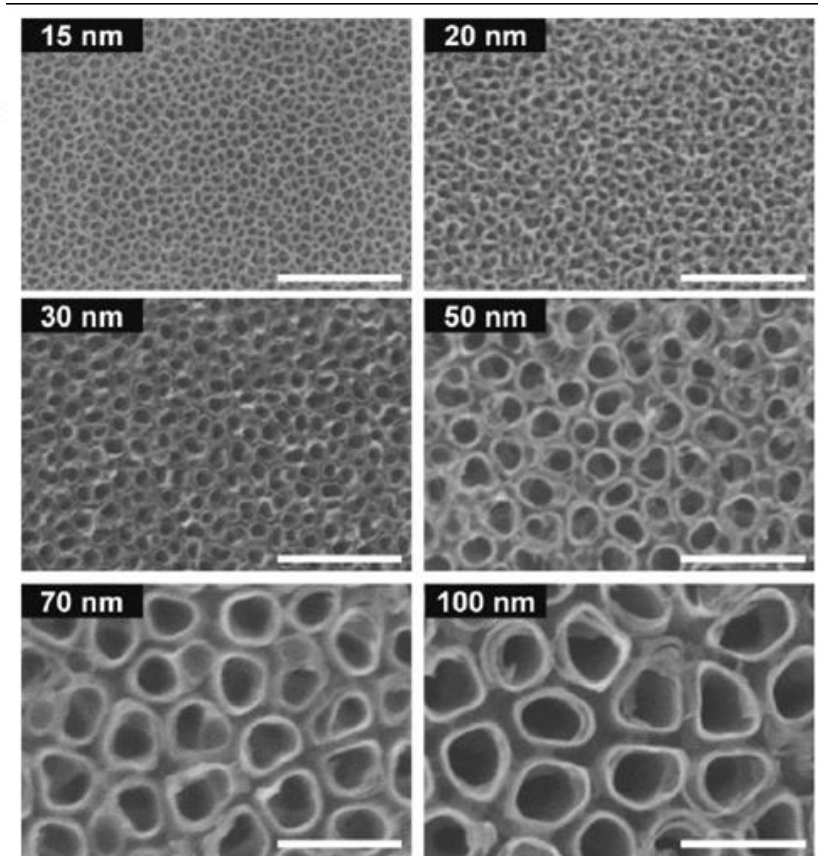
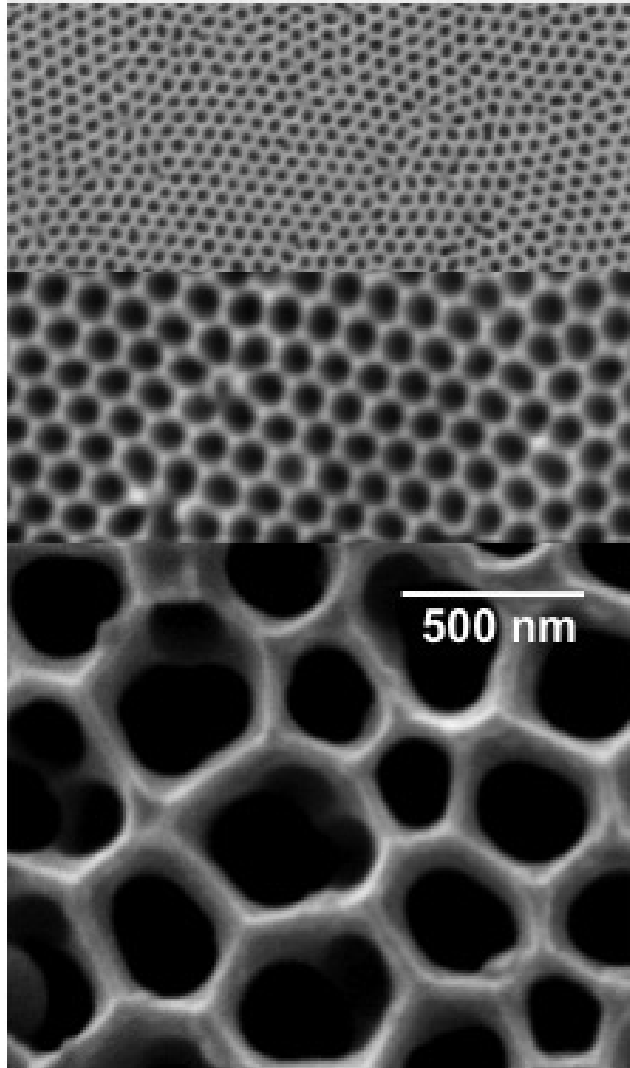


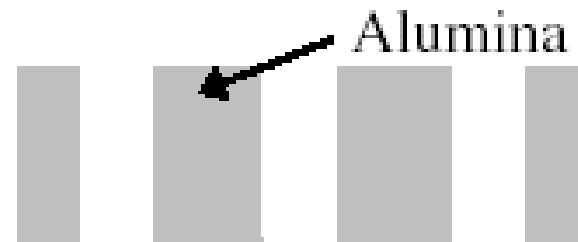
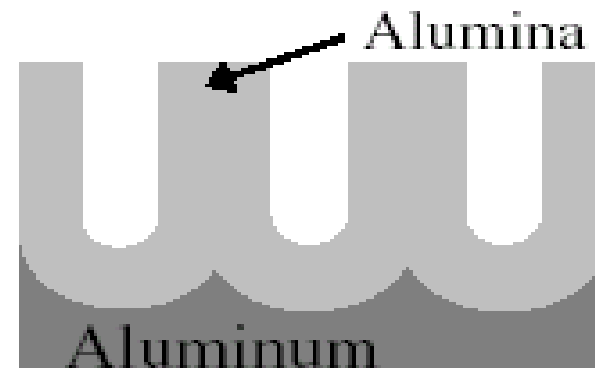
Figure 1. Top-view SEM images of self-assembled layers of vertically oriented TiO₂ nanotubes of six different diameters ranging between 15 and 100 nm formed in 1 M H₃PO₄ + 0.3 wt% HF at potentials between 1 and 20 V for 1 h. Scale bars: 200 nm.



Membrane PAA - PAT



Ossidazione anodica di
Alluminio 99,99X %





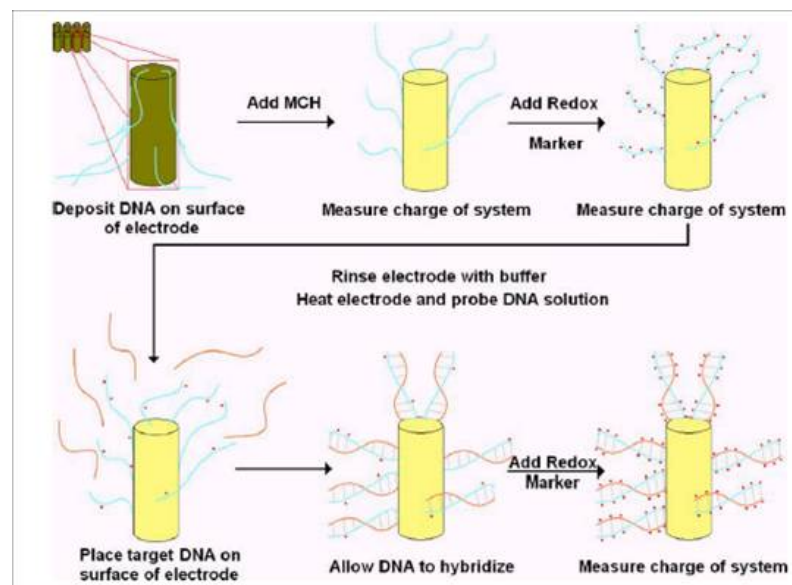
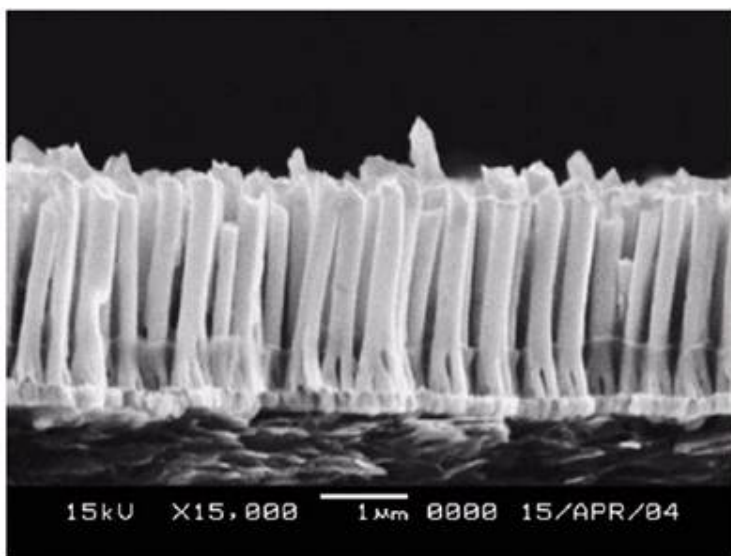
Detection of DNA oligonucleotides on nanowire array electrodes using chronocoulometry

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J. Tres Brazell, Mahnaz El-Kouedi*

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Received 25 March 2005; received in revised form 16 July 2005; accepted 30 July 2005

Available online 21 October 2005





A novel miniature gas ionization sensor based on freestanding gold nanowires

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Received 28 November 2006; received in revised form 8 March 2007; accepted 10 March 2007

Available online 15 March 2007

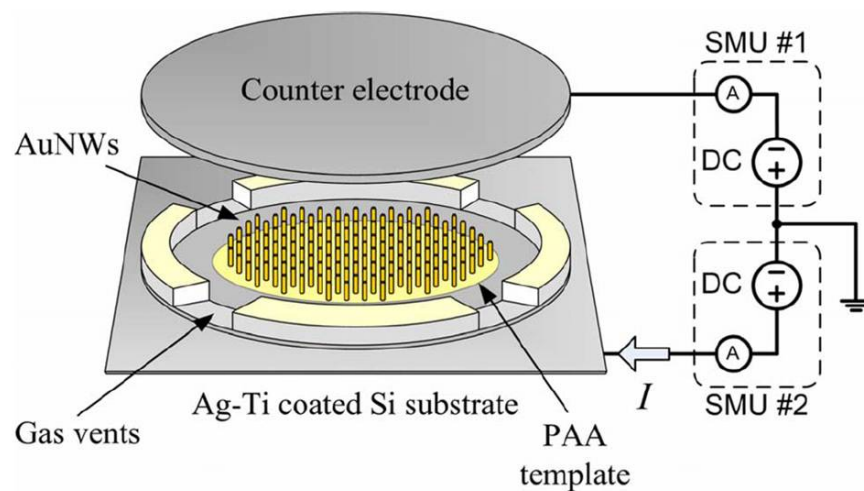


Fig. 6. Schematic illustration of the GFIS cell and the measurement setup. The counter electrode is mounted on the peripheral supporting ring. I denotes the field-ion current.



PAA per Biosensori



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Microporous and Mesoporous Materials 111 (2008) 359–366

MICROPOROUS AND
MESOPOROUS MATERIALS

www.elsevier.com/locate/micromeso

Study on the activity and stability of urease immobilized onto nanoporous alumina membranes

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Received 26 July 2006; received in revised form 22 November 2006; accepted 10 August 2007

Available online 17 August 2007

Abstract

Nanoporous alumina membranes were employed as substrate materials for urease immobilization. Anodic porous alumina was prepared by the two-step anodization of high purity aluminum. By controlling anodization conditions, the nanoporous structure with desired dimension was obtained. Urease immobilization onto nanoporous alumina membranes was performed by four different protocols. Effect of pore diameter, pore length and immobilization methods on the activity and stability of immobilized enzyme was discussed in detail. The results show that the enzymes immobilized onto porous alumina with big pore diameter possess high activity and poor stability as compared to small pore diameter. The effect of pore length is complicated, the activity of enzyme increases with the increasing pore length for big pore size; while for correspondingly small pore size, enzymatic activity slightly depends on pore length. The immobilization methods have a slight effect on enzymatic activity, whereas enzyme immobilization by chitosan coating and reticulation with glutaraldehyde exhibits a good long-term stability as compared to that only via physical adsorption.

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Keywords: Urease; Immobilization; Activity; Stability; Nanoporous alumina membranes

Catalytic nanoliths

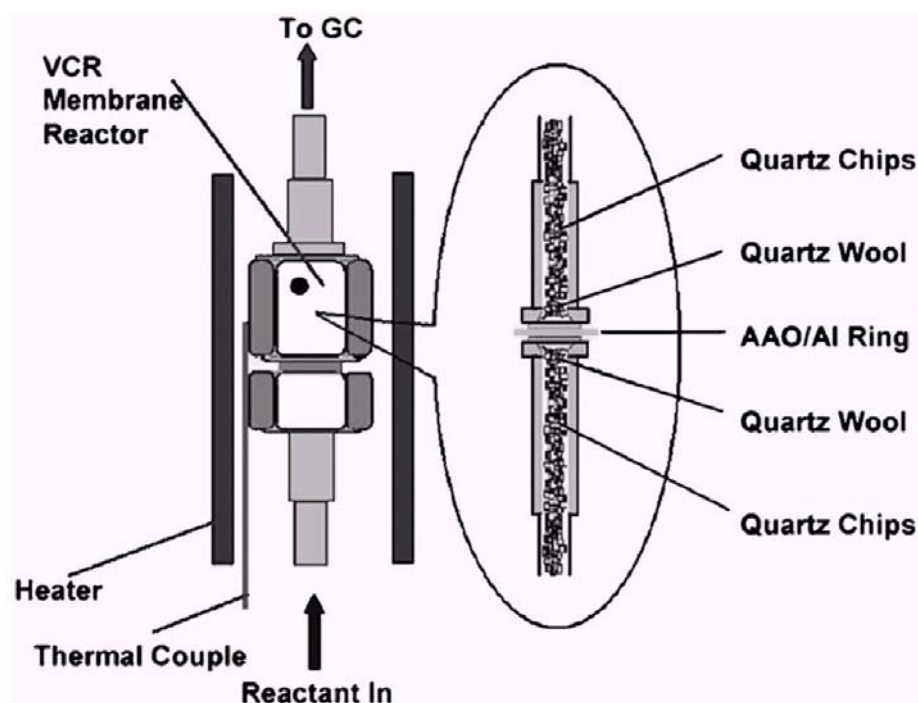
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ABSTRACT: The nanoporous anodic aluminum oxide (AAO) structure is shown to be a useful platform for heterogeneous catalysis. By appropriately masking the perimeter during anodization and etching, the AAO can be formed at the center of an aluminum disc. The remaining aluminum ring connects seamlessly to the AAO and provides mechanical support for convenient handling. The supported AAO can be sealed in a standard fitting so that the nanopores in the structure function as an array of tubular reactors, i.e. a nanolith. Coating the walls with catalytically active materials turns the nanolith into a novel catalytic system.

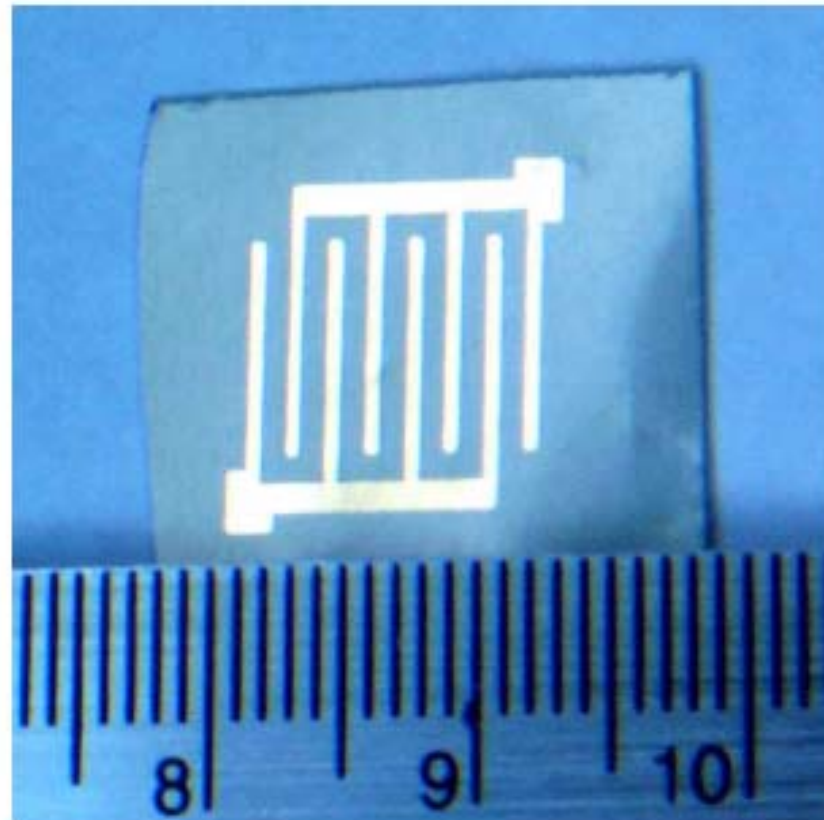
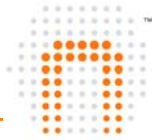


Figure 1. Digital image of a typical nanoporous alumina sensor.



Metallic Nanowires & Nanoparticles



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A novel miniature gas ionization sensor based on freestanding gold nanowires

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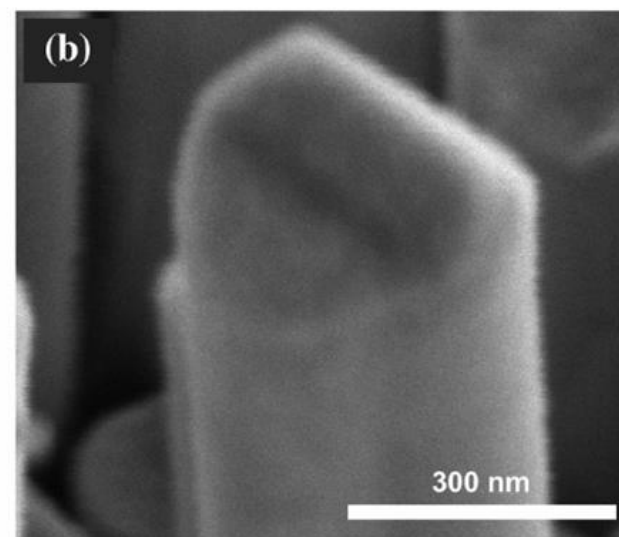
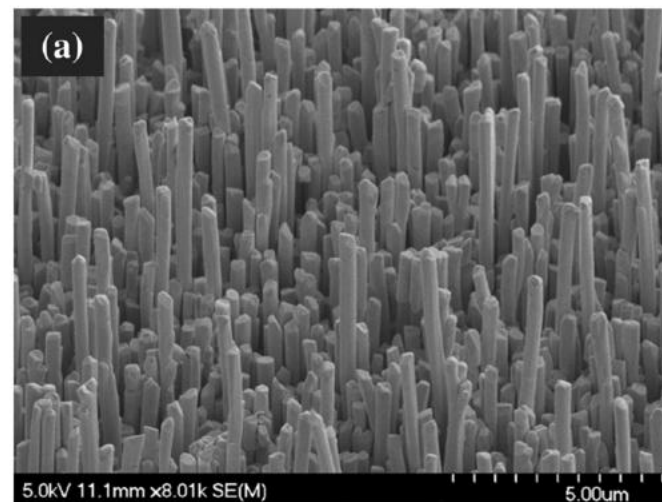
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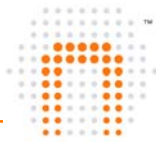
Abstract

Freestanding gold nanowires (AuNW) were incorporated to fabricate a miniaturized gas ionization sensor. The device exhibited improved sensitivity in sub-Torr pressure (P) regime compared to similar devices reported earlier, since the room temperature breakdown voltage (V_b) was further reduced in low gas pressures or concentrations (N). Also excellent selectivity was achieved for pressures up to 10 Torr, while V_b remained almost unaffected by pressure. In the V_b - P characteristic below 1 Torr, V_b was less when the AuNWs was configured as cathode, and it started to decline until a Paschen-like (V_b)_{min} was observed at very low pressures ($0.3 < P < 0.5$ Torr). The reduction of breakdown voltage in this case, was attributed to the creation of an abnormally large amount of secondary electrons at the nanowire tips due to the increase in Townsend's secondary ionization factor (ω/α) at very high reduced fields (E/N). The pre-breakdown I - V characteristic of the device, obtained at extreme low pressures showed polarity dependency. The field enhancement factor (β) of the AuNW array was estimated by operating the device in the ohmic discharge region. The average aspect-ratio of the nanowires, which was extracted from the I - V characteristic, showed good agreement with SEM observations.

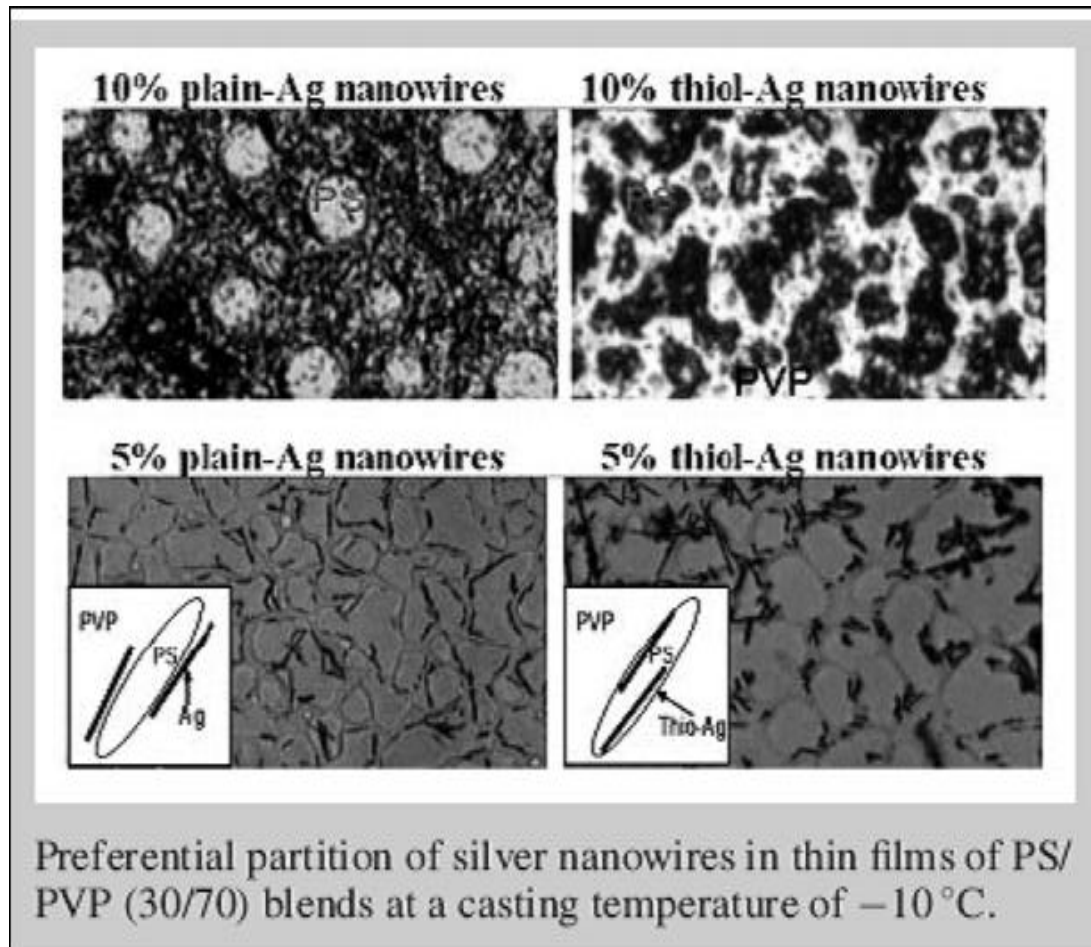
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Keywords: Gas ionization sensor; Gold nanowires; Breakdown voltage; Field enhancement; Sub-Torr pressure





Materiali nanocompositi: Matrice polimerica + nanofili-nanoparticelle



Electrical Conductivity in Polymer Composites Containing Metal Nanowires: Simulation and Experiment

[White, Sadie](#); [Vemulkar, Tarun](#); [Fischer, John](#); [Winey, Karen](#)

American Physical Society, 2009 APS March Meeting, March 16-20, 2009, abstract #P19.008

The study of rod percolation behavior has resurfaced in recent years, because it explains electrical conductivity in polymer nanocomposites containing carbon nanotubes and metal nanowires. Common processing techniques result in fillers with $L/D < 50$, so traditional models, which are only strictly correct in the limit of $L/D \sim \infty$, are ineffective at predicting percolation in these systems. We present a simulation that constructs percolated networks of finite-aspect ratio rods and calculates their electrical conductivity. We will compare our simulation results with polymer composites containing silver nanowires with aspect ratios of ~ 10 and ~ 30 . Finally, we will present the temperature-dependent electrical conductivity of these composites and interpret the results using the thermal expansion coefficients of polystyrene and silver. These materials act as "thermal switches," wherein electrical conductivity of certain composites can be manipulated by several orders of magnitude over the temperature range from 80K-425 K.

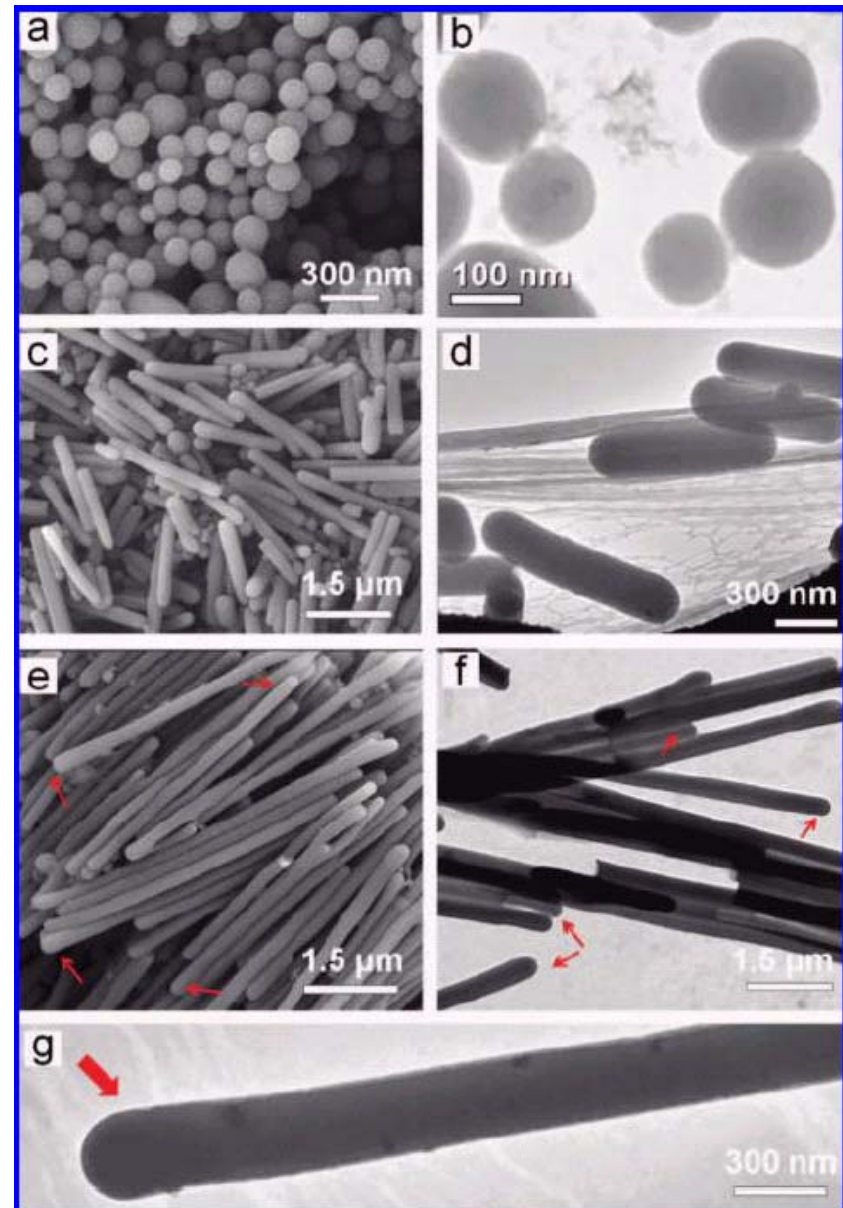
Polymer Composites with Oriented Magnetic Nanowires as Fillers

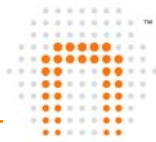
Li Sun, Kusuma Keshoju

Metallic nickel nanowires with excellent physical properties have been introduced into polydimethylsiloxane matrix to form polymer nanocomposites. Nanowires were synthesized by template-assisted electrochemical deposition. By utilizing ferromagnetic nickel nanowires, small external magnetic field can be used to control their alignment and distribution during composite synthesis. Unlike dielectrophoresis, optical tweezers, and microfluidic flow control, magnetic manipulation provides a cost-effective, non-contact, and versatile approach to control nanostructured materials in fluids over a large area. Polydimethylsiloxane composites with nanowires arranged in longitudinal, transverse, and random orientations with respect to the applied load direction were studied. Tensile tests showed that the composites with longitudinal arrangement have higher elastic modulus and tensile strength than the other composite samples. Experimentally obtained elastic modulus values were compared with the prediction of classical Halpin-Tsai model. Metallic nickel nanowires with excellent physical properties have been introduced into polydimethylsiloxane matrix to form polymer nanocomposites. Nanowires were synthesized by template-assisted electrochemical deposition. By utilizing ferromagnetic nickel nanowires, small external magnetic field can be used to control their alignment and distribution during composite synthesis. Unlike dielectrophoresis, optical tweezers, and microfluidic flow control, magnetic manipulation provides a cost-effective, non-contact, and versatile approach to control nanostructured materials in fluids over a large area. Polydimethylsiloxane composites with nanowires arranged in longitudinal, transverse, and random orientations with respect to the applied load direction were studied. Tensile tests showed that the composites with longitudinal arrangement have higher elastic modulus and tensile strength than the other composite samples. Experimentally obtained elastic modulus values were compared with the prediction of classical Halpin-Tsai model.

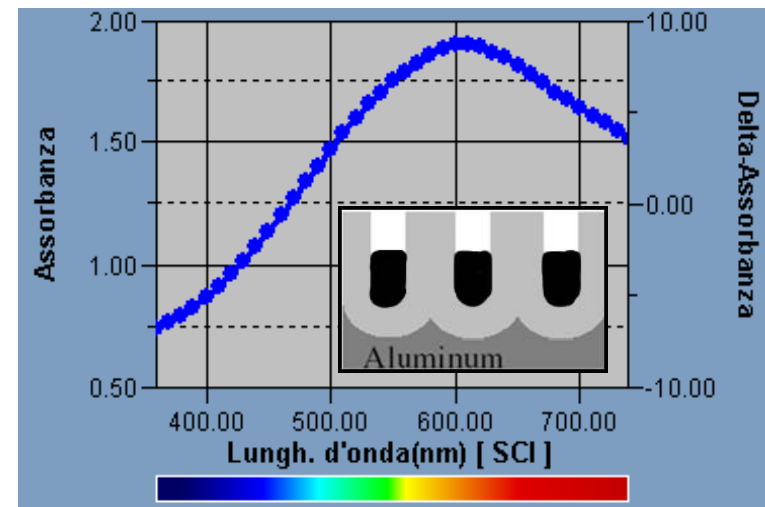
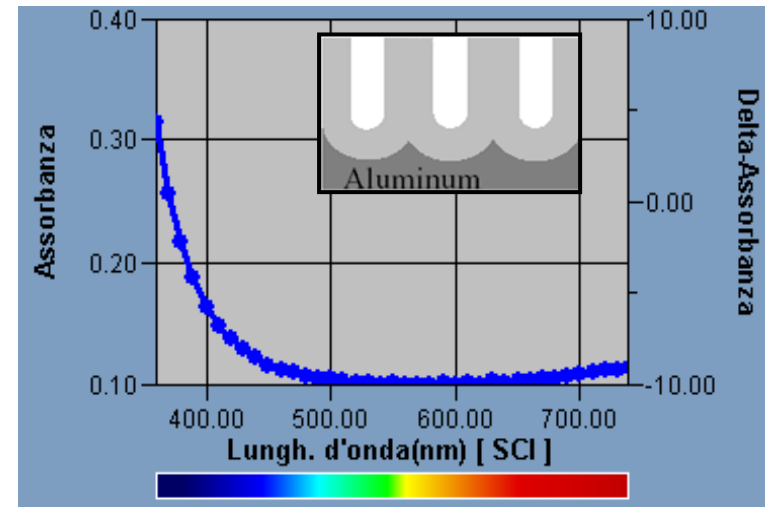
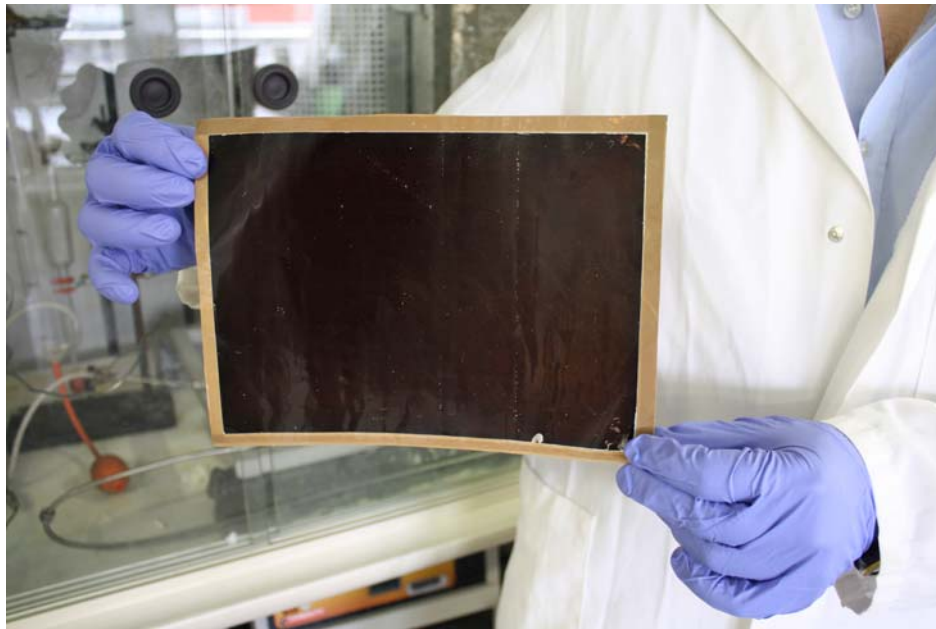


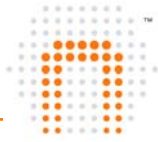
- Nanoparticles
- Nanorods
- Nanowires



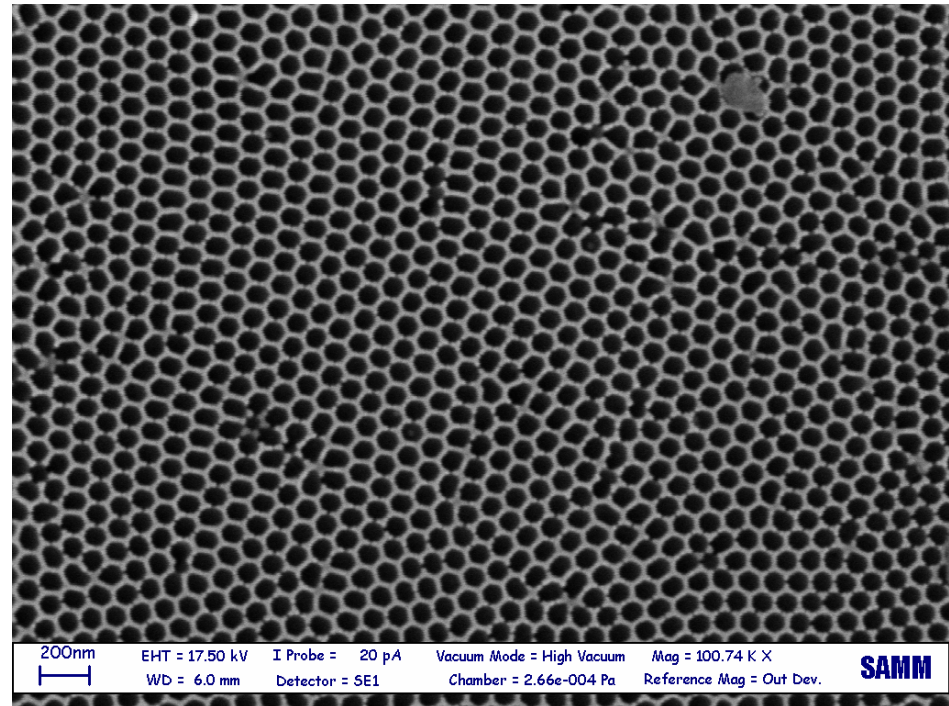
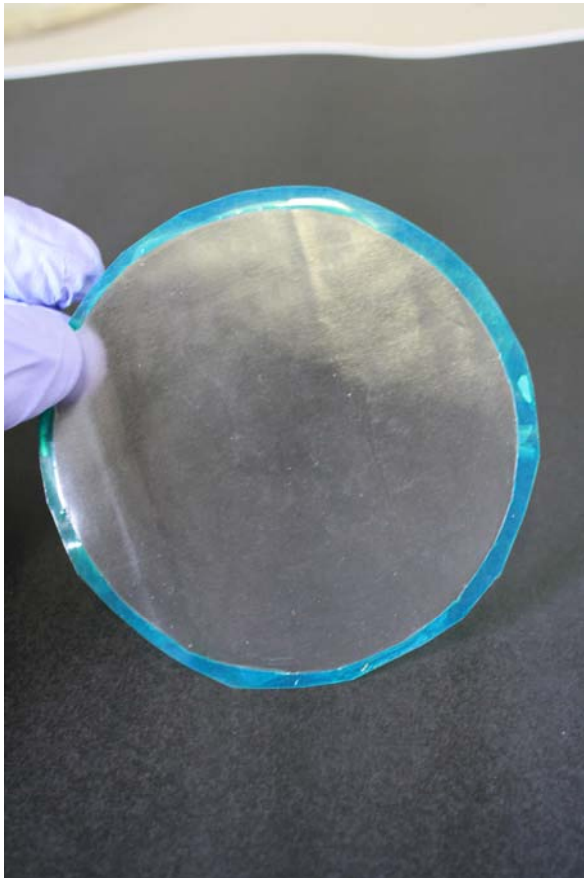


Supporti nanostrutturati per Assorbitori Solari



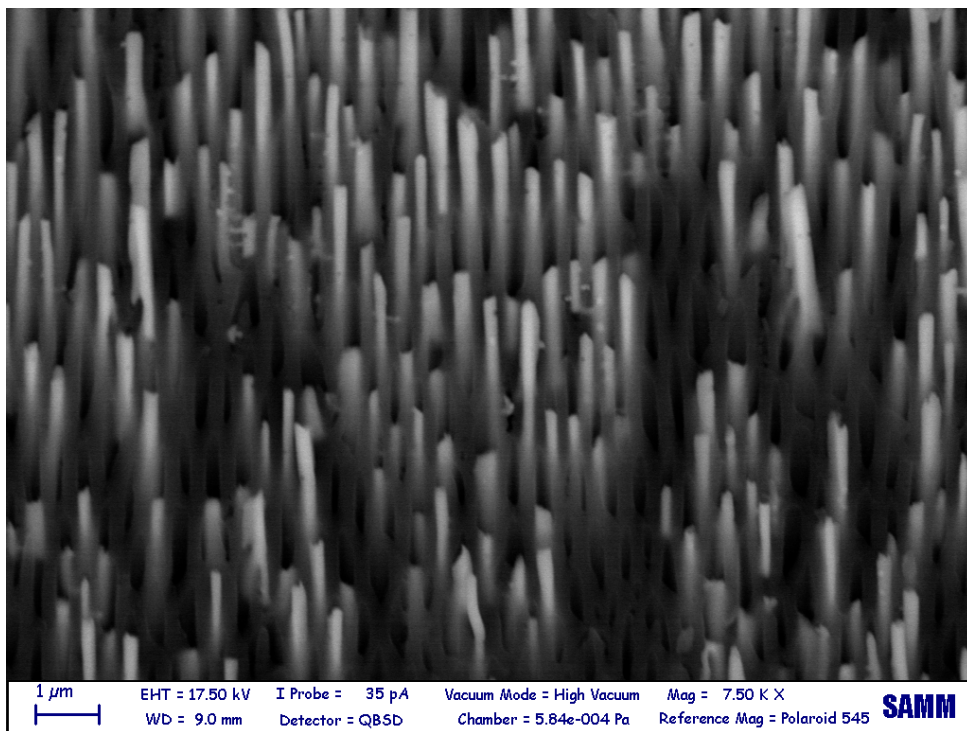


Membrana PAA Ø 90 mm





Cu NWs in PPA



Cu NWs in PPG

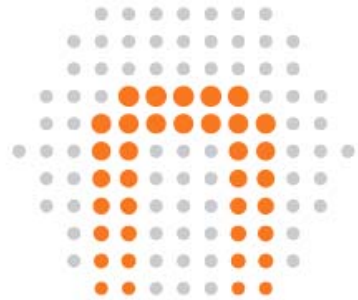




Soluzioni Ag Nanoparticles



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4 → 20 nm



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